

# WHO AM I?

These mystery elements are waiting to be identified. The trick is—you'll need the Periodic Table to unmask their identities. Unless you have it memorized, you'll need a copy of the table from your science book or from page 119 of this book. Read the clues about each mystery element, figure out what it is, and then write the name and symbol of the element.

**REMEMBER:**

The Atomic number equals the number of protons. Atomic mass equals the number of protons plus neutrons. The number of electrons equals the number of protons.

1. \_\_\_\_\_ 11. \_\_\_\_\_

2. \_\_\_\_\_ 12. \_\_\_\_\_

3. \_\_\_\_\_ 13. \_\_\_\_\_

4. \_\_\_\_\_ 14. \_\_\_\_\_

5. \_\_\_\_\_ 15. \_\_\_\_\_

6. \_\_\_\_\_ 16. \_\_\_\_\_

7. \_\_\_\_\_ 17. \_\_\_\_\_

8. \_\_\_\_\_ 18. \_\_\_\_\_

9. \_\_\_\_\_ 19. \_\_\_\_\_

10. \_\_\_\_\_ 20. \_\_\_\_\_

5. nonmetallic period 3 atomic mass 32  
6. 26 protons period 4 transition element  
8. 29 electrons period 4  
9. atomic mass 20 gas  
10. period 5 transition element 51 neutrons  
11. 80 electrons transition element  
12. lowest mass in period  
13. metallic period 4 20 electrons  
14. period 6 gas 86 protons  
15. 4 neutrons metallic  
16. period 4 metallic 27 electrons  
17. metallic period 6 56 protons  
18. gas atomic mass 16 8 neutrons  
19. mass less than 30 not neon noble gas  
20. period 5 metallic 38 electrons

Name \_\_\_\_\_